

Trifluoroacetic acid, 4-chlorobenzyl ester

Other names:	4-Chlorobenzyl alcohol, trifluoroacetate
Inchi:	InChI=1S/C9H6ClF3O2/c10-7-3-1-6(2-4-7)5-15-8(14)9(11,12)13/h1-4H,5H2
InchiKey:	WRYXVYBIODRCTM-UHFFFAOYSA-N
Formula:	C9H6ClF3O2
SMILES:	O=C(OCc1ccc(Cl)cc1)C(F)(F)F
Mol. weight [g/mol]:	238.59
CAS:	38696-07-0

Physical Properties

Property code	Value	Unit	Source
gf	-699.76	kJ/mol	Joback Method
hf	-861.65	kJ/mol	Joback Method
hfus	21.53	kJ/mol	Joback Method
hvap	48.36	kJ/mol	Joback Method
log10ws	-3.40		Crippen Method
logp	2.946		Crippen Method
mcvol	138.900	ml/mol	McGowan Method
pc	2850.52	kPa	Joback Method
rinpol	1184.00		NIST Webbook
rinpol	1184.00		NIST Webbook
tb	545.28	K	Joback Method
tc	748.34	K	Joback Method
tf	336.40	K	Joback Method
vc	0.547	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	309.43	J/molxK	545.28	Joback Method
cpg	319.98	J/molxK	579.12	Joback Method
cpg	329.81	J/molxK	612.97	Joback Method
cpg	338.95	J/molxK	646.81	Joback Method
cpg	347.43	J/molxK	680.65	Joback Method
cpg	355.28	J/molxK	714.49	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C38696070&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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