

# Naphthalene, 1,3,6,8-tetranitro-

<b>Other names:</b>	1,3,6,8-Tetranitronaphthalene
<b>Inchi:</b>	InChI=1S/C10H4N4O8/c15-11(16)6-1-5-2-7(12(17)18)4-9(14(21)22)10(5)8(3-6)13(19)20
<b>InchiKey:</b>	BFMBQYIQQLBGJA-UHFFFAOYSA-N
<b>Formula:</b>	C10H4N4O8
<b>SMILES:</b>	O=[N+]([O-])c1cc([N+](=O)[O-])c2c([N+](=O)[O-])cc([N+](=O)[O-])cc2c1
<b>Mol. weight [g/mol]:</b>	308.16
<b>CAS:</b>	28995-89-3

## Physical Properties

Property code	Value	Unit	Source
gf	356.06	kJ/mol	Joback Method
hf	88.95	kJ/mol	Joback Method
hfus	56.60	kJ/mol	Joback Method
hvap	110.78	kJ/mol	Joback Method
log10ws	-5.88		Crippen Method
logp	2.473		Crippen Method
mcvol	178.220	ml/mol	McGowan Method
pc	3838.78	kPa	Joback Method
rinpol	442.85		NIST Webbook
rinpol	442.85		NIST Webbook
tb	1101.14	K	Joback Method
tc	1409.82	K	Joback Method
tf	886.10	K	Joback Method
vc	0.738	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	512.06	J/molxK	1101.14	Joback Method
cpg	518.05	J/molxK	1152.59	Joback Method
cpg	523.72	J/molxK	1204.03	Joback Method
cpg	529.21	J/molxK	1255.48	Joback Method
cpg	534.69	J/molxK	1306.93	Joback Method
cpg	540.30	J/molxK	1358.38	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C28995893&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C28995893&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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