

P-phenylenediamine, n,n-dimethyl, n',n'-bis(p-tolylsulfonyl)-

Inchi:	InChI=1S/C22H24N2O4S2/c1-17-5-13-21(14-6-17)29(25,26)24(20-11-9-19(10-12-20)23
InchiKey:	CXIBFOYRMRNMRY-UHFFFAOYSA-N
Formula:	C22H24N2O4S2
SMILES:	Cc1ccc(S(=O)(=O)N(c2ccc(N(C)C)cc2)S(=O)(=O)c2ccc(C)cc2)cc1
Mol. weight [g/mol]:	444.57
CAS:	19770-85-5

Physical Properties

Property code	Value	Unit	Source
gf	-272.82	kJ/mol	Joback Method
hf	-593.87	kJ/mol	Joback Method
hfus	62.49	kJ/mol	Joback Method
hvap	114.74	kJ/mol	Joback Method
log10ws	-5.23		Crippen Method
logp	3.954		Crippen Method
mcvol	325.700	ml/mol	McGowan Method
pc	2320.31	kPa	Joback Method
tb	918.18	K	Joback Method
tc	1146.79	K	Joback Method
tf	596.58	K	Joback Method
vc	1.232	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	977.64	J/molxK	918.18	Joback Method
cpg	990.95	J/molxK	956.28	Joback Method
cpg	1002.58	J/molxK	994.38	Joback Method
cpg	1012.57	J/molxK	1032.49	Joback Method
cpg	1021.00	J/molxK	1070.59	Joback Method
cpg	1027.94	J/molxK	1108.69	Joback Method
cpg	1033.46	J/molxK	1146.79	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C19770855&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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