

# 4-Nitro-4'-aminodiphenylsulfone

<b>Other names:</b>	4-(4-Nitrophenylsulfonyl)aniline Benzenamine, 4-[(4-nitrophenyl)sulfonyl]-
<b>Inchi:</b>	InChI=1S/C12H10N2O4S/c13-9-1-5-11(6-2-9)19(17,18)12-7-3-10(4-8-12)14(15)16/h1-8H
<b>InchiKey:</b>	DMZVYFFBWHBWMO-UHFFFAOYSA-N
<b>Formula:</b>	C12H10N2O4S
<b>SMILES:</b>	<chem>Nc1ccc(S(=O)(=O)c2ccc([N+](=O)[O-])cc2)cc1</chem>
<b>Mol. weight [g/mol]:</b>	278.28
<b>CAS:</b>	1948-92-1

## Physical Properties

Property code	Value	Unit	Source
gf	-110.82	kJ/mol	Joback Method
hf	-271.21	kJ/mol	Joback Method
hfus	42.08	kJ/mol	Joback Method
hvap	94.05	kJ/mol	Joback Method
log10ws	-3.12		Crippen Method
logp	2.010		Crippen Method
mcvol	187.910	ml/mol	McGowan Method
pc	4362.63	kPa	Joback Method
tb	809.43	K	Joback Method
tc	1073.94	K	Joback Method
tf	568.31	K	Joback Method
vc	0.729	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	508.76	J/molxK	809.43	Joback Method
cpg	519.78	J/molxK	853.51	Joback Method
cpg	529.41	J/molxK	897.60	Joback Method
cpg	537.71	J/molxK	941.68	Joback Method
cpg	544.72	J/molxK	985.77	Joback Method
cpg	550.50	J/molxK	1029.85	Joback Method
cpg	555.11	J/molxK	1073.94	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1948921&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1948921&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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