

# 2-(2-Trifluoroacetoxy-5-methoxy-4-ethylphenyl)ethyl

Other names:

4-ethyl-2-(1-dimethoxy-«beta»-phenethylamine-M, (O-desmethyl), isomer 1, 2TFA

**N-trifluoroacetyl-**

InChI=1S/C15H15F6NO4/c1-3-8-6-11(26-13(24)15(19,20)21)9(7-10(8)25-2)4-5-22-12(23)

**InchiKey:** DEDBLQUFJTWCQT-UHFFFAOYSA-N

**Formula:** C15H15F6NO4

**SMILES:** CCc1cc(OC(=O)C(F)(F)F)c(CCNC(=O)C(F)(F)F)cc1OC

**Mol. weight [g/mol]:** 387.27

## Physical Properties

Property code	Value	Unit	Source
gf	-1382.69	kJ/mol	Joback Method
hf	-1781.10	kJ/mol	Joback Method
hfus	41.80	kJ/mol	Joback Method
hvap	70.50	kJ/mol	Joback Method
log10ws	-4.64		Crippen Method
logp	2.946		Crippen Method
mcvol	233.930	ml/mol	McGowan Method
pc	1601.28	kPa	Joback Method
rinpol	1740.00		NIST Webbook
rinpol	1740.00		NIST Webbook
tb	776.13	K	Joback Method
tc	963.18	K	Joback Method
tf	528.15	K	Joback Method
vc	0.936	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	696.59	J/molxK	776.13	Joback Method
cpg	708.18	J/molxK	807.30	Joback Method
cpg	718.94	J/molxK	838.48	Joback Method
cpg	728.92	J/molxK	869.65	Joback Method
cpg	738.14	J/molxK	900.83	Joback Method
cpg	746.64	J/molxK	932.00	Joback Method
cpg	754.45	J/molxK	963.18	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U360353&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U360353&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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