

# Sebacic acid, dodecyl pentafluorophenyl ester

**Inchi:** InChI=1S/C28H41F5O4/c1-2-3-4-5-6-7-8-11-14-17-20-36-21(34)18-15-12-9-10-13-16-19  
**InchiKey:** UBZNGALJNXLVKQ-UHFFFAOYSA-N  
**Formula:** C28H41F5O4  
**SMILES:** CCCCCCCCCCOC(=O)CCCCCCCCC(=O)Oc1c(F)c(F)c(F)c(F)c1F  
**Mol. weight [g/mol]:** 536.61

## Physical Properties

Property code	Value	Unit	Source
gf	-1192.75	kJ/mol	Joback Method
hf	-1912.22	kJ/mol	Joback Method
hfus	81.35	kJ/mol	Joback Method
hvap	97.73	kJ/mol	Joback Method
log10ws	-10.68		Crippen Method
logp	8.872		Crippen Method
mvol	405.350	ml/mol	McGowan Method
pc	698.76	kPa	Joback Method
rinpol	3060.00		NIST Webbook
rinpol	3060.00		NIST Webbook
tb	1040.55	K	Joback Method
tc	1305.98	K	Joback Method
tf	641.61	K	Joback Method
vc	1.633	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1399.67	J/molxK	1040.55	Joback Method
cpg	1418.18	J/molxK	1084.79	Joback Method
cpg	1434.27	J/molxK	1129.03	Joback Method
cpg	1448.01	J/molxK	1173.27	Joback Method
cpg	1459.46	J/molxK	1217.51	Joback Method
cpg	1468.70	J/molxK	1261.74	Joback Method
cpg	1475.77	J/molxK	1305.98	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U355034&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U355034&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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