

Benzamide, 3-trifluoromethyl-N-ethyl-N-propyl-

Inchi:	InChI=1S/C13H16F3NO/c1-3-8-17(4-2)12(18)10-6-5-7-11(9-10)13(14,15)16/h5-7,9H,3-4
InchiKey:	LWVMRQVYLJIFOQ-UHFFFAOYSA-N
Formula:	C13H16F3NO
SMILES:	CCCN(CC)C(=O)c1cccc(C(F)(F)F)c1
Mol. weight [g/mol]:	259.27

Physical Properties

Property code	Value	Unit	Source
gf	-438.37	kJ/mol	Joback Method
hf	-728.72	kJ/mol	Joback Method
hfus	29.52	kJ/mol	Joback Method
hvap	52.51	kJ/mol	Joback Method
log10ws	-3.97		Crippen Method
logp	3.578		Crippen Method
mcvol	187.130	ml/mol	McGowan Method
pc	2056.76	kPa	Joback Method
rinpol	2706.00		NIST Webbook
rinpol	2706.00		NIST Webbook
tb	589.39	K	Joback Method
tc	776.96	K	Joback Method
tf	361.80	K	Joback Method
vc	0.723	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	481.18	J/mol×K	589.39	Joback Method
cpg	496.18	J/mol×K	620.65	Joback Method
cpg	510.25	J/mol×K	651.91	Joback Method
cpg	523.44	J/mol×K	683.18	Joback Method
cpg	535.79	J/mol×K	714.44	Joback Method
cpg	547.35	J/mol×K	745.70	Joback Method
cpg	558.17	J/mol×K	776.96	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U415568&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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