

# Glutaric acid, 3-methylbut-2-yl 4-chloro-2-methoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C17H23ClO5/c1-11(2)12(3)22-16(19)6-5-7-17(20)23-14-9-8-13(18)10-15(14)2
<b>InchiKey:</b>	ISFZYCWVHUVITK-UHFFFAOYSA-N
<b>Formula:</b>	C17H23ClO5
<b>SMILES:</b>	COc1cc(Cl)ccc1OC(=O)CCCC(=O)OC(C)C(C)C
<b>Mol. weight [g/mol]:</b>	342.81

## Physical Properties

Property code	Value	Unit	Source
gf	-404.24	kJ/mol	Joback Method
hf	-828.74	kJ/mol	Joback Method
hfus	36.96	kJ/mol	Joback Method
hvap	81.37	kJ/mol	Joback Method
log10ws	-4.67		Crippen Method
logp	4.012		Crippen Method
mcvol	259.620	ml/mol	McGowan Method
pc	1606.42	kPa	Joback Method
rinpola	2341.00		NIST Webbook
rinpola	2341.00		NIST Webbook
tb	836.55	K	Joback Method
tc	1046.56	K	Joback Method
tf	499.28	K	Joback Method
vc	0.983	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	759.12	J/molxK	836.55	Joback Method
cpg	817.62	J/molxK	1011.56	Joback Method
cpg	808.25	J/molxK	976.56	Joback Method
cpg	797.72	J/molxK	941.56	Joback Method
cpg	786.02	J/molxK	906.55	Joback Method
cpg	773.15	J/molxK	871.55	Joback Method
cpg	825.81	J/molxK	1046.56	Joback Method
dvisc	0.0000462	Paxs	836.55	Joback Method

dvisc	0.0000596	Paxs	780.34	Joback Method
dvisc	0.0000799	Paxs	724.13	Joback Method
dvisc	0.0001125	Paxs	667.91	Joback Method
dvisc	0.0001688	Paxs	611.70	Joback Method
dvisc	0.0002748	Paxs	555.49	Joback Method
dvisc	0.0004995	Paxs	499.28	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U393903&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U393903&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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