

Isophthalic acid, diamide, N,N'-dibutyl-

Inchi:	InChI=1S/C16H24N2O2/c1-3-5-10-17-15(19)13-8-7-9-14(12-13)16(20)18-11-6-4-2/h7-9,
InchiKey:	UEUJQJBBZXOQII-UHFFFAOYSA-N
Formula:	C16H24N2O2
SMILES:	CCCCNC(=O)c1cccc(C(=O)NCCCC)c1
Mol. weight [g/mol]:	276.37

Physical Properties

Property code	Value	Unit	Source
gf	107.56	kJ/mol	Joback Method
hf	-266.73	kJ/mol	Joback Method
hfus	44.24	kJ/mol	Joback Method
hvap	80.51	kJ/mol	Joback Method
log10ws	-4.67		Crippen Method
logp	2.746		Crippen Method
mvol	235.640	ml/mol	McGowan Method
pc	1913.58	kPa	Joback Method
rinpol	2746.00		NIST Webbook
rinpol	2746.00		NIST Webbook
tb	805.22	K	Joback Method
tc	1011.68	K	Joback Method
tf	514.20	K	Joback Method
vc	0.905	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	705.19	J/mol×K	805.22	Joback Method
cpg	719.66	J/mol×K	839.63	Joback Method
cpg	733.17	J/mol×K	874.04	Joback Method
cpg	745.75	J/mol×K	908.45	Joback Method
cpg	757.44	J/mol×K	942.86	Joback Method
cpg	768.30	J/mol×K	977.27	Joback Method
cpg	778.37	J/mol×K	1011.68	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U345813&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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