

3-Chloro-2-fluorobenzoic acid, hexyl ester

Inchi:	InChI=1S/C13H16ClFO2/c1-2-3-4-5-9-17-13(16)10-7-6-8-11(14)12(10)15/h6-8H,2-5,9H2
InchiKey:	YFRSPDVBDUHKAP-UHFFFAOYSA-N
Formula:	C13H16ClFO2
SMILES:	CCCCCCOC(=O)c1cccc(Cl)c1F
Mol. weight [g/mol]:	258.72

Physical Properties

Property code	Value	Unit	Source
gf	-288.93	kJ/mol	Joback Method
hf	-554.71	kJ/mol	Joback Method
hfus	32.75	kJ/mol	Joback Method
hvap	60.86	kJ/mol	Joback Method
log10ws	-4.82		Crippen Method
logp	4.216		Crippen Method
mvol	191.720	ml/mol	McGowan Method
pc	2064.24	kPa	Joback Method
rinpol	1817.00		NIST Webbook
rinpol	1817.00		NIST Webbook
tb	646.47	K	Joback Method
tc	846.27	K	Joback Method
tf	390.40	K	Joback Method
vc	0.747	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	478.71	J/molxK	646.47	Joback Method
cpg	492.49	J/molxK	679.77	Joback Method
cpg	505.51	J/molxK	713.07	Joback Method
cpg	517.78	J/molxK	746.37	Joback Method
cpg	529.32	J/molxK	779.67	Joback Method
cpg	540.13	J/molxK	812.97	Joback Method
cpg	550.25	J/molxK	846.27	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U338883&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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