

# Succinic acid, 2,4,6-trichlorophenyl 2-methoxyethyl ester

<b>Inchi:</b>	InChI=1S/C13H13Cl3O5/c1-19-4-5-20-11(17)2-3-12(18)21-13-9(15)6-8(14)7-10(13)16/h
<b>InchiKey:</b>	UOXUVKZZXJWCPX-UHFFFAOYSA-N
<b>Formula:</b>	C13H13Cl3O5
<b>SMILES:</b>	COCCOC(=O)CCC(=O)Oc1c(Cl)cc(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	355.60

## Physical Properties

Property code	Value	Unit	Source
gf	-466.53	kJ/mol	Joback Method
hf	-778.57	kJ/mol	Joback Method
hfus	41.65	kJ/mol	Joback Method
hvap	82.67	kJ/mol	Joback Method
log10ws	-3.88		Crippen Method
logp	3.522		Crippen Method
mvol	227.740	ml/mol	McGowan Method
pc	2036.39	kPa	Joback Method
rinpol	2320.00		NIST Webbook
rinpol	2320.00		NIST Webbook
tb	825.75	K	Joback Method
tc	1044.21	K	Joback Method
tf	556.56	K	Joback Method
vc	0.869	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	580.66	J/molxK	825.75	Joback Method
cpg	590.88	J/molxK	862.16	Joback Method
cpg	600.14	J/molxK	898.57	Joback Method
cpg	608.42	J/molxK	934.98	Joback Method
cpg	615.70	J/molxK	971.39	Joback Method
cpg	621.97	J/molxK	1007.80	Joback Method
cpg	627.21	J/molxK	1044.21	Joback Method
dvisc	0.0003737	Paxs	556.56	Joback Method

dvisc	0.0002572	Paxs	601.43	Joback Method
dvisc	0.0001865	Paxs	646.29	Joback Method
dvisc	0.0001409	Paxs	691.15	Joback Method
dvisc	0.0001102	Paxs	736.02	Joback Method
dvisc	0.0000887	Paxs	780.88	Joback Method
dvisc	0.0000731	Paxs	825.75	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390749&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390749&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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