

2,6-Difluorobenzoic acid, 3,5-dimethylphenyl ester

Inchi:	InChI=1S/C15H12F2O2/c1-9-6-10(2)8-11(7-9)19-15(18)14-12(16)4-3-5-13(14)17/h3-8H,
InchiKey:	ODMQMIUUJGEWTQ-UHFFFAOYSA-N
Formula:	C15H12F2O2
SMILES:	Cc1cc(C)cc(OC(=O)c2c(F)cccc2F)c1
Mol. weight [g/mol]:	262.25

Physical Properties

Property code	Value	Unit	Source
gf	-361.82	kJ/mol	Joback Method
hf	-562.77	kJ/mol	Joback Method
hfus	30.08	kJ/mol	Joback Method
hvap	63.71	kJ/mol	Joback Method
log10ws	-5.17		Crippen Method
logp	3.801		Crippen Method
mvol	185.670	ml/mol	McGowan Method
pc	2267.57	kPa	Joback Method
rinpol	1796.00		NIST Webbook
rinpol	1796.00		NIST Webbook
tb	690.71	K	Joback Method
tc	911.11	K	Joback Method
tf	435.07	K	Joback Method
vc	0.720	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	478.74	J/mol×K	690.71	Joback Method
cpg	492.36	J/mol×K	727.44	Joback Method
cpg	505.05	J/mol×K	764.18	Joback Method
cpg	516.83	J/mol×K	800.91	Joback Method
cpg	527.72	J/mol×K	837.64	Joback Method
cpg	537.73	J/mol×K	874.38	Joback Method
cpg	546.89	J/mol×K	911.11	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U307558&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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