

# Diethylmalonic acid, 2,2,3,3,3-pentafluoropropyl tetradecyl ester

Inchi:	InChI=1S/C24H41F5O4/c1-4-7-8-9-10-11-12-13-14-15-16-17-18-32-20(30)22(5-2,6-3)21
InchiKey:	GHMSDZJTPNHTHE-UHFFFAOYSA-N
Formula:	C24H41F5O4
SMILES:	CCCCCCCCCCCCCOC(=O)C(CC)(CC)C(=O)OCC(F)(F)C(F)(F)F
Mol. weight [g/mol]:	488.57

## Physical Properties

Property code	Value	Unit	Source
gf	-1282.17	kJ/mol	Joback Method
hf	-2035.09	kJ/mol	Joback Method
hfus	56.65	kJ/mol	Joback Method
hvap	79.36	kJ/mol	Joback Method
log10ws	-8.33		Crippen Method
logp	7.778		Crippen Method
mvol	372.750	ml/mol	McGowan Method
pc	772.46	kPa	Joback Method
rinpol	2205.00		NIST Webbook
rinpol	2205.00		NIST Webbook
tb	887.76	K	Joback Method
tc	1089.87	K	Joback Method
tf	514.77	K	Joback Method
vc	1.484	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1253.40	J/molxK	887.76	Joback Method
cpg	1272.47	J/molxK	921.44	Joback Method
cpg	1290.30	J/molxK	955.13	Joback Method
cpg	1307.00	J/molxK	988.81	Joback Method
cpg	1322.65	J/molxK	1022.50	Joback Method
cpg	1337.34	J/molxK	1056.18	Joback Method
cpg	1351.16	J/molxK	1089.87	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370851&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370851&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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