

# Succinic acid, 5-fluoro-2-nitrophenyl pentafluorobenzyl ester

<b>Inchi:</b>	InChI=1S/C17H9F6NO6/c18-7-1-2-9(24(27)28)10(5-7)30-12(26)4-3-11(25)29-6-8-13(19)
<b>InchiKey:</b>	VZSGBCZDEHFGTP-UHFFFAOYSA-N
<b>Formula:</b>	C17H9F6NO6
<b>SMILES:</b>	O=C(CCC(=O)Oc1cc(F)ccc1[N+](=O)[O-])OCc1c(F)c(F)c(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	437.25

## Physical Properties

Property code	Value	Unit	Source
gf	-1351.48	kJ/mol	Joback Method
hf	-1678.46	kJ/mol	Joback Method
hfus	60.56	kJ/mol	Joback Method
hvap	92.62	kJ/mol	Joback Method
log10ws	-6.65		Crippen Method
logp	3.858		Crippen Method
mvol	245.790	ml/mol	McGowan Method
pc	1653.80	kPa	Joback Method
rinpol	2464.00		NIST Webbook
rinpol	2464.00		NIST Webbook
tb	976.62	K	Joback Method
tc	1199.72	K	Joback Method
tf	713.30	K	Joback Method
vc	1.010	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	737.06	J/molxK	976.62	Joback Method
cpg	744.27	J/molxK	1013.80	Joback Method
cpg	750.29	J/molxK	1050.99	Joback Method
cpg	755.11	J/molxK	1088.17	Joback Method
cpg	758.74	J/molxK	1125.35	Joback Method
cpg	761.17	J/molxK	1162.54	Joback Method
cpg	762.39	J/molxK	1199.72	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357982&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357982&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rlnol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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