

# Benzoic acid, 4-nitro-, anhydride

<b>Other names:</b>	p-Nitrobenzoic acid anhydride
<b>Inchi:</b>	InChI=1S/C14H8N2O7/c17-13(9-1-5-11(6-2-9)15(19)20)23-14(18)10-3-7-12(8-4-10)16(2
<b>InchiKey:</b>	JYMVSZGJZRQOFY-UHFFFAOYSA-N
<b>Formula:</b>	C14H8N2O7
<b>SMILES:</b>	O=C(OC(=O)c1ccc([N+](=O)[O-])cc1)c1ccc([N+](=O)[O-])cc1
<b>Mol. weight [g/mol]:</b>	316.22
<b>CAS:</b>	902-47-6

## Physical Properties

Property code	Value	Unit	Source
gf	-19.18	kJ/mol	Joback Method
hf	-261.07	kJ/mol	Joback Method
hfus	46.43	kJ/mol	Joback Method
hvap	101.72	kJ/mol	Joback Method
log10ws	-4.98		Crippen Method
logp	2.500		Crippen Method
mcvol	204.450	ml/mol	McGowan Method
pc	3096.73	kPa	Joback Method
tb	1016.88	K	Joback Method
tc	1296.07	K	Joback Method
tf	734.73	K	Joback Method
vc	0.797	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	590.04	J/mol×K	1016.88	Joback Method
cpg	596.26	J/mol×K	1063.41	Joback Method
cpg	601.26	J/mol×K	1109.94	Joback Method
cpg	605.12	J/mol×K	1156.48	Joback Method
cpg	607.91	J/mol×K	1203.01	Joback Method
cpg	609.71	J/mol×K	1249.54	Joback Method
cpg	610.59	J/mol×K	1296.07	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C902476&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C902476&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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