

Glutaric acid, butyl 2-iodobenzyl ester

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|-----------------------------|---|
| Inchi: | InChI=1S/C16H21IO4/c1-2-3-11-20-15(18)9-6-10-16(19)21-12-13-7-4-5-8-14(13)17/h4-5 |
| InchiKey: | TUHKBOZMEUAYBZ-UHFFFAOYSA-N |
| Formula: | C16H21IO4 |
| SMILES: | CCCCOC(=O)CCCC(=O)OCc1ccccc1I |
| Mol. weight [g/mol]: | 404.24 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -223.10 | kJ/mol | Joback Method |
| hf | -561.24 | kJ/mol | Joback Method |
| hfus | 40.83 | kJ/mol | Joback Method |
| hvap | 81.83 | kJ/mol | Joback Method |
| log10ws | -4.99 | | Crippen Method |
| logp | 3.848 | | Crippen Method |
| mvol | 253.240 | ml/mol | McGowan Method |
| pc | 1783.35 | kPa | Joback Method |
| rinpol | 2468.00 | | NIST Webbook |
| rinpol | 2468.00 | | NIST Webbook |
| tb | 842.86 | K | Joback Method |
| tc | 1064.56 | K | Joback Method |
| tf | 511.40 | K | Joback Method |
| vc | 0.960 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 698.57 | J/molxK | 842.86 | Joback Method |
| cpg | 754.13 | J/molxK | 1027.61 | Joback Method |
| cpg | 745.04 | J/molxK | 990.66 | Joback Method |
| cpg | 734.96 | J/molxK | 953.71 | Joback Method |
| cpg | 723.88 | J/molxK | 916.76 | Joback Method |
| cpg | 711.76 | J/molxK | 879.81 | Joback Method |
| cpg | 762.27 | J/molxK | 1064.56 | Joback Method |
| dvisc | 0.0000704 | Paxs | 842.86 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0000895 | Paxs | 787.62 | Joback Method |
| dvisc | 0.0001181 | Paxs | 732.37 | Joback Method |
| dvisc | 0.0001629 | Paxs | 677.13 | Joback Method |
| dvisc | 0.0002381 | Paxs | 621.89 | Joback Method |
| dvisc | 0.0003746 | Paxs | 566.64 | Joback Method |
| dvisc | 0.0006500 | Paxs | 511.40 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U376878&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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<https://www.chemeo.com/cid/122-425-4/Glutaric-acid-butyl-2-iodobenzyl-ester.pdf>

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