

Glutaric acid, hexyl 1-(pentafluorophenyl)ethyl ester

Inchi:	InChI=1S/C19H23F5O4/c1-3-4-5-6-10-27-12(25)8-7-9-13(26)28-11(2)14-15(20)17(22)19
InchiKey:	QYROSCFOAVAICX-UHFFFAOYSA-N
Formula:	C19H23F5O4
SMILES:	CCCCCOC(=O)CCCC(=O)OC(C)c1c(F)c(F)c(F)c(F)c1F
Mol. weight [g/mol]:	410.38

Physical Properties

Property code	Value	Unit	Source
gf	-1270.97	kJ/mol	Joback Method
hf	-1731.74	kJ/mol	Joback Method
hfus	54.51	kJ/mol	Joback Method
hvap	77.31	kJ/mol	Joback Method
log10ws	-6.72		Crippen Method
logp	5.280		Crippen Method
mcvol	278.540	ml/mol	McGowan Method
pc	1180.09	kPa	Joback Method
rinpol	2124.00		NIST Webbook
rinpol	2124.00		NIST Webbook
tb	834.19	K	Joback Method
tc	1022.89	K	Joback Method
tf	525.18	K	Joback Method
vc	1.123	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	853.81	J/molxK	834.19	Joback Method
cpg	867.81	J/molxK	865.64	Joback Method
cpg	880.85	J/molxK	897.09	Joback Method
cpg	892.92	J/molxK	928.54	Joback Method
cpg	904.02	J/molxK	959.99	Joback Method
cpg	914.16	J/molxK	991.44	Joback Method
cpg	923.35	J/molxK	1022.89	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U377000&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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