

# Trichloroacetic acid, decyl ester

<b>Other names:</b>	Decyl trichloroacetate
<b>Inchi:</b>	InChI=1S/C12H21Cl3O2/c1-2-3-4-5-6-7-8-9-10-17-11(16)12(13,14)15/h2-10H2,1H3
<b>InchiKey:</b>	JVOWHDCGJSEUMO-UHFFFAOYSA-N
<b>Formula:</b>	C12H21Cl3O2
<b>SMILES:</b>	CCCCCCCCCOC(=O)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	303.65
<b>CAS:</b>	65611-33-8

## Physical Properties

Property code	Value	Unit	Source
gf	-216.71	kJ/mol	Joback Method
hf	-591.78	kJ/mol	Joback Method
hfus	34.80	kJ/mol	Joback Method
hvap	63.32	kJ/mol	Joback Method
log10ws	-5.27		Crippen Method
logp	5.041		Crippen Method
mcvol	224.100	ml/mol	McGowan Method
pc	1713.19	kPa	Joback Method
rinpol	1760.00		NIST Webbook
rinpol	1762.70		NIST Webbook
rinpol	1760.00		NIST Webbook
rinpol	1762.70		NIST Webbook
rinpol	1771.00		NIST Webbook
ripol	2160.00		NIST Webbook
ripol	2160.00		NIST Webbook
tb	659.31	K	Joback Method
tc	851.10	K	Joback Method
tf	389.34	K	Joback Method
vc	0.868	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	558.87	J/mol×K	659.31	Joback Method

cpg	572.87	J/molxK	691.27	Joback Method
cpg	586.07	J/molxK	723.24	Joback Method
cpg	598.52	J/molxK	755.20	Joback Method
cpg	610.24	J/molxK	787.17	Joback Method
cpg	621.27	J/molxK	819.13	Joback Method
cpg	631.64	J/molxK	851.10	Joback Method
dvisc	0.0018663	Paxs	389.34	Joback Method
dvisc	0.0009436	Paxs	434.33	Joback Method
dvisc	0.0005423	Paxs	479.33	Joback Method
dvisc	0.0003427	Paxs	524.32	Joback Method
dvisc	0.0002329	Paxs	569.32	Joback Method
dvisc	0.0001675	Paxs	614.31	Joback Method
dvisc	0.0001260	Paxs	659.31	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C65611338&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C65611338&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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