

# octyl 3,5-dinitrobenzoate

<b>Other names:</b>	Benzoic acid, 3,5-dinitro, octyl ester
<b>Inchi:</b>	InChI=1S/C15H20N2O6/c1-2-3-4-5-6-7-8-23-15(18)12-9-13(16(19)20)11-14(10-12)17(21)
<b>InchiKey:</b>	UUJMMZMVYHKYGJ-UHFFFAOYSA-N
<b>Formula:</b>	C15H20N2O6
<b>SMILES:</b>	CCCCCCCCOC(=O)c1cc([N+](=O)[O-])cc([N+](=O)[O-])c1
<b>Mol. weight [g/mol]:</b>	324.33
<b>CAS:</b>	10478-06-5

## Physical Properties

Property code	Value	Unit	Source
gf	5.75	kJ/mol	Joback Method
hf	-405.66	kJ/mol	Joback Method
hfus	53.38	kJ/mol	Joback Method
hvap	94.92	kJ/mol	Joback Method
log10ws	-5.95		Crippen Method
logp	4.020		Crippen Method
mcvol	240.730	ml/mol	McGowan Method
pc	1911.91	kPa	Joback Method
rinpol	2343.00		NIST Webbook
rinpol	2343.00		NIST Webbook
rinpol	2355.00		NIST Webbook
rinpol	2368.00		NIST Webbook
rinpol	2380.00		NIST Webbook
rinpol	2343.00		NIST Webbook
rinpol	2370.00		NIST Webbook
rinpol	2370.00		NIST Webbook
ripol	3265.00		NIST Webbook
ripol	3288.00		NIST Webbook
ripol	3305.00		NIST Webbook
ripol	3313.00		NIST Webbook
ripol	3313.00		NIST Webbook
ripol	3265.00		NIST Webbook
ripol	3265.00		NIST Webbook
tb	959.21	K	Joback Method
tc	1197.77	K	Joback Method
tf	669.65	K	Joback Method
vc	0.956	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	757.28	J/mol×K	959.21	Joback Method
cpg	768.30	J/mol×K	998.97	Joback Method
cpg	778.18	J/mol×K	1038.73	Joback Method
cpg	786.96	J/mol×K	1078.49	Joback Method
cpg	794.69	J/mol×K	1118.25	Joback Method
cpg	801.42	J/mol×K	1158.01	Joback Method
cpg	807.19	J/mol×K	1197.77	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10478065&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10478065&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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