

3-Methoxy-2,4,5-trifluorobenzoic acid, 2,2,2-trichloroethyl ester

Inchi:	InChI=1S/C10H6Cl3F3O3/c1-18-8-6(15)4(2-5(14)7(8)16)9(17)19-3-10(11,12)13/h2H,3H2
InchiKey:	ZQBRAJZNGINPAK-UHFFFAOYSA-N
Formula:	C10H6Cl3F3O3
SMILES:	COc1c(F)c(F)cc(C(=O)OCC(Cl)(Cl)Cl)c1F
Mol. weight [g/mol]:	337.51

Physical Properties

Property code	Value	Unit	Source
gf	-849.09	kJ/mol	Joback Method
hf	-1080.40	kJ/mol	Joback Method
hfus	32.53	kJ/mol	Joback Method
hvap	63.75	kJ/mol	Joback Method
log10ws	-4.81		Crippen Method
logp	3.640		Crippen Method
mvol	183.340	ml/mol	McGowan Method
pc	2256.81	kPa	Joback Method
rinpol	1739.00		NIST Webbook
rinpol	1739.00		NIST Webbook
tb	680.38	K	Joback Method
tc	888.91	K	Joback Method
tf	467.30	K	Joback Method
vc	0.720	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	419.14	J/mol×K	680.38	Joback Method
cpg	428.03	J/mol×K	715.13	Joback Method
cpg	436.30	J/mol×K	749.89	Joback Method
cpg	443.95	J/mol×K	784.64	Joback Method
cpg	450.99	J/mol×K	819.40	Joback Method
cpg	457.43	J/mol×K	854.15	Joback Method
cpg	463.27	J/mol×K	888.91	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U360572&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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