

# Ethiofencarb

<b>Other names:</b>	Phenol, 2-[(ethylthio)methyl]-, methylcarbamate Arylmate Croneton 2-[(Ethylsulfanyl)methyl]phenyl methylcarbamate Carbamic acid, methyl-, «alpha»-(ethylthio)-o-tolyl ester CHOX 1901 Croneton 500 HOX 1901 «alpha»-Ethylthio-o-tolyl methylcarbamate
<b>Inchi:</b>	InChI=1S/C11H15NO2S/c1-3-15-8-9-6-4-5-7-10(9)14-11(13)12-2/h4-7H,3,8H2,1-2H3,(H
<b>InchiKey:</b>	HEZNVIIYQEUHLNI-UHFFFAOYSA-N
<b>Formula:</b>	C11H15NO2S
<b>SMILES:</b>	CCSCc1cccc1OC(=O)NC
<b>Mol. weight [g/mol]:</b>	225.31
<b>CAS:</b>	56729-20-5

## Physical Properties

Property code	Value	Unit	Source
gf	33.11	kJ/mol	Joback Method
hf	-194.77	kJ/mol	Joback Method
hfus	29.91	kJ/mol	Joback Method
hvap	65.43	kJ/mol	Joback Method
log10ws	-3.55		Crippen Method
logp	2.658		Crippen Method
mcvol	175.860	ml/mol	McGowan Method
pc	2823.32	kPa	Joback Method
rinpol	1780.00		NIST Webbook
rinpol	1780.00		NIST Webbook
rinpol	1780.00		NIST Webbook
rinpol	1788.00		NIST Webbook
rinpol	1824.00		NIST Webbook
ripol	2378.00		NIST Webbook
ripol	2378.00		NIST Webbook
tb	677.98	K	Joback Method
tc	904.24	K	Joback Method
tf	411.89	K	Joback Method
vc	0.656	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	448.20	J/mol×K	677.98	Joback Method
cpg	461.98	J/mol×K	715.69	Joback Method
cpg	474.81	J/mol×K	753.40	Joback Method
cpg	486.70	J/mol×K	791.11	Joback Method
cpg	497.68	J/mol×K	828.82	Joback Method
cpg	507.75	J/mol×K	866.53	Joback Method
cpg	516.92	J/mol×K	904.24	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C56729205&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C56729205&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/122-806-1/Ethiofencarb.pdf>

Generated by Cheméo on 2024-05-02 02:02:01.254832136 +0000 UTC m=+16904570.175409451.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.