

# 2-Bromobenzoic acid, phenyl ester

<b>Inchi:</b>	InChI=1S/C13H9BrO2/c14-12-9-5-4-8-11(12)13(15)16-10-6-2-1-3-7-10/h1-9H
<b>InchiKey:</b>	SNOWYPDEJVKXAG-UHFFFAOYSA-N
<b>Formula:</b>	C13H9BrO2
<b>SMILES:</b>	O=C(Oc1ccccc1)c1ccccc1Br
<b>Mol. weight [g/mol]:</b>	277.11

## Physical Properties

Property code	Value	Unit	Source
gf	54.17	kJ/mol	Joback Method
hf	-68.53	kJ/mol	Joback Method
hfus	25.19	kJ/mol	Joback Method
hvap	65.34	kJ/mol	Joback Method
log10ws	-4.72		Crippen Method
logp	3.668		Crippen Method
mcvol	171.450	ml/mol	McGowan Method
pc	3443.98	kPa	Joback Method
rinpol	1894.00		NIST Webbook
rinpol	1894.00		NIST Webbook
tb	697.63	K	Joback Method
tc	958.04	K	Joback Method
tf	433.59	K	Joback Method
vc	0.633	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	399.78	J/molxK	697.63	Joback Method
cpg	412.50	J/molxK	741.03	Joback Method
cpg	424.05	J/molxK	784.43	Joback Method
cpg	434.51	J/molxK	827.84	Joback Method
cpg	443.93	J/molxK	871.24	Joback Method
cpg	452.40	J/molxK	914.64	Joback Method
cpg	459.97	J/molxK	958.04	Joback Method
dvisc	0.0010570	Paxs	433.59	Joback Method

dvisc	0.0006618	Paxs	477.60	Joback Method
dvisc	0.0004484	Paxs	521.60	Joback Method
dvisc	0.0003228	Paxs	565.61	Joback Method
dvisc	0.0002437	Paxs	609.62	Joback Method
dvisc	0.0001911	Paxs	653.62	Joback Method
dvisc	0.0001545	Paxs	697.63	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U299015&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U299015&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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