

# 2,4-dichlorobenzyl hexyl ether

**Inchi:** InChI=1S/C13H18Cl2O/c1-2-3-4-5-8-16-10-11-6-7-12(14)9-13(11)15/h6-7,9H,2-5,8,10H2  
**InchiKey:** RTCLSTOXGULHFW-UHFFFAOYSA-N  
**Formula:** C13H18Cl2O  
**SMILES:** CCCCCOCc1ccc(Cl)cc1Cl  
**Mol. weight [g/mol]:** 261.19

## Physical Properties

Property code	Value	Unit	Source
gf	22.87	kJ/mol	Joback Method
hf	-261.76	kJ/mol	Joback Method
hfus	32.27	kJ/mol	Joback Method
hvap	59.31	kJ/mol	Joback Method
log10ws	-5.32		Crippen Method
logp	5.090		Crippen Method
mcvol	200.620	ml/mol	McGowan Method
pc	1964.82	kPa	Joback Method
rinpol	1768.00		NIST Webbook
rinpol	1759.00		NIST Webbook
rinpol	1759.00		NIST Webbook
rinpol	1764.00		NIST Webbook
rinpol	1766.00		NIST Webbook
rinpol	1765.00		NIST Webbook
rinpol	1762.00		NIST Webbook
rinpol	1765.00		NIST Webbook
rinpol	1769.00		NIST Webbook
rinpol	1770.00		NIST Webbook
rinpol	1758.00		NIST Webbook
rinpol	1763.00		NIST Webbook
rinpol	1765.00		NIST Webbook
rinpol	1767.00		NIST Webbook
tb	630.76	K	Joback Method
tc	836.05	K	Joback Method
tf	369.80	K	Joback Method
vc	0.771	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	479.97	J/molxK	630.76	Joback Method
cpg	546.29	J/molxK	801.83	Joback Method
cpg	534.57	J/molxK	767.62	Joback Method
cpg	522.10	J/molxK	733.40	Joback Method
cpg	508.85	J/molxK	699.19	Joback Method
cpg	494.82	J/molxK	664.97	Joback Method
cpg	557.29	J/molxK	836.05	Joback Method
dvisc	0.0001400	Paxs	630.76	Joback Method
dvisc	0.0001756	Paxs	587.27	Joback Method
dvisc	0.0002284	Paxs	543.77	Joback Method
dvisc	0.0003110	Paxs	500.28	Joback Method
dvisc	0.0004491	Paxs	456.79	Joback Method
dvisc	0.0007006	Paxs	413.29	Joback Method
dvisc	0.0012134	Paxs	369.80	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R32307&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R32307&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/122-914-1/2-4-dichlorobenzyl-hexyl-ether.pdf>

Generated by Cheméo on 2024-04-27 15:47:52.282595188 +0000 UTC m=+16522121.203172501.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.