

# Succinic acid, cyclohexylmethyl 3-chlorophenyl ester

<b>Inchi:</b>	InChI=1S/C17H21ClO4/c18-14-7-4-8-15(11-14)22-17(20)10-9-16(19)21-12-13-5-2-1-3-6
<b>InchiKey:</b>	RQLJLLQTXFWGOJ-UHFFFAOYSA-N
<b>Formula:</b>	C17H21ClO4
<b>SMILES:</b>	O=C(CCC(=O)Oc1cccc(Cl)c1)OCC1CCCCC1
<b>Mol. weight [g/mol]:</b>	324.80

## Physical Properties

Property code	Value	Unit	Source
gf	-260.28	kJ/mol	Joback Method
hf	-620.17	kJ/mol	Joback Method
hfus	35.04	kJ/mol	Joback Method
hvap	79.50	kJ/mol	Joback Method
log10ws	-4.75		Crippen Method
logp	4.149		Crippen Method
mvol	242.890	ml/mol	McGowan Method
pc	1933.83	kPa	Joback Method
rinpol	2475.00		NIST Webbook
rinpol	2475.00		NIST Webbook
tb	829.58	K	Joback Method
tc	1058.62	K	Joback Method
tf	501.91	K	Joback Method
vc	0.909	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	725.06	J/molxK	829.58	Joback Method
cpg	740.53	J/molxK	867.75	Joback Method
cpg	754.56	J/molxK	905.93	Joback Method
cpg	767.18	J/molxK	944.10	Joback Method
cpg	778.43	J/molxK	982.27	Joback Method
cpg	788.33	J/molxK	1020.44	Joback Method
cpg	796.90	J/molxK	1058.62	Joback Method
dvisc	0.0007495	Paxs	501.91	Joback Method

dvisc	0.0004230	Paxs	556.52	Joback Method
dvisc	0.0002645	Paxs	611.13	Joback Method
dvisc	0.0001786	Paxs	665.75	Joback Method
dvisc	0.0001280	Paxs	720.36	Joback Method
dvisc	0.0000961	Paxs	774.97	Joback Method
dvisc	0.0000750	Paxs	829.58	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389854&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389854&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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