

Benzenesulfonamide, 4-methyl-N-ethyl-N-pentyl-

Inchi:	InChI=1S/C14H23NO2S/c1-4-6-7-12-15(5-2)18(16,17)14-10-8-13(3)9-11-14/h8-11H,4-7,
InchiKey:	FUNQQIFBSVLTMW-UHFFFAOYSA-N
Formula:	C14H23NO2S
SMILES:	CCCCCN(CC)S(=O)(=O)c1ccc(C)cc1
Mol. weight [g/mol]:	269.40

Physical Properties

Property code	Value	Unit	Source
gf	-187.98	kJ/mol	Joback Method
hf	-493.05	kJ/mol	Joback Method
hfus	40.07	kJ/mol	Joback Method
hvap	70.37	kJ/mol	Joback Method
log10ws	-3.71		Crippen Method
logp	3.196		Crippen Method
mvol	222.430	ml/mol	McGowan Method
pc	2282.77	kPa	Joback Method
rinpol	2238.00		NIST Webbook
rinpol	2238.00		NIST Webbook
tb	611.60	K	Joback Method
tc	799.14	K	Joback Method
tf	357.51	K	Joback Method
vc	0.856	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	570.20	J/mol×K	611.60	Joback Method
cpg	587.93	J/mol×K	642.86	Joback Method
cpg	604.68	J/mol×K	674.11	Joback Method
cpg	620.49	J/mol×K	705.37	Joback Method
cpg	635.37	J/mol×K	736.63	Joback Method
cpg	649.34	J/mol×K	767.88	Joback Method
cpg	662.44	J/mol×K	799.14	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U415275&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rlnol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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