

Succinic acid, 2-octyl 1,1,1-trifluoro-2-propyl ester

Inchi:	InChI=1S/C15H25F3O4/c1-4-5-6-7-8-11(2)21-13(19)9-10-14(20)22-12(3)15(16,17)18/h1
InchiKey:	UCXZMHZHCQTLEA-UHFFFAOYSA-N
Formula:	C15H25F3O4
SMILES:	CCCCCCC(C)OC(=O)CCC(=O)OC(C)C(F)(F)F
Mol. weight [g/mol]:	326.35

Physical Properties

Property code	Value	Unit	Source
gf	-978.89	kJ/mol	Joback Method
hf	-1450.17	kJ/mol	Joback Method
hfus	34.96	kJ/mol	Joback Method
hvap	62.77	kJ/mol	Joback Method
log10ws	-4.71		Crippen Method
logp	4.163		Crippen Method
mvol	242.400	ml/mol	McGowan Method
pc	1406.95	kPa	Joback Method
rinpol	1558.00		NIST Webbook
rinpol	1558.00		NIST Webbook
tb	688.88	K	Joback Method
tc	859.96	K	Joback Method
tf	377.32	K	Joback Method
vc	0.955	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	704.11	J/mol×K	688.88	Joback Method
cpg	719.49	J/mol×K	717.39	Joback Method
cpg	734.07	J/mol×K	745.91	Joback Method
cpg	747.89	J/mol×K	774.42	Joback Method
cpg	760.95	J/mol×K	802.94	Joback Method
cpg	773.27	J/mol×K	831.45	Joback Method
cpg	784.87	J/mol×K	859.96	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U370972&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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