

# Succinic acid, 2,3-dichlorophenyl 2,4-dimethylpent-3-yl ester

**Inchi:** InChI=1S/C17H22Cl2O4/c1-10(2)17(11(3)4)23-15(21)9-8-14(20)22-13-7-5-6-12(18)16(19)  
**InchiKey:** TXBCZXVXAZCKBL-UHFFFAOYSA-N  
**Formula:** C17H22Cl2O4  
**SMILES:** CC(C)C(OC(=O)CCC(=O)Oc1cccc(Cl)c1Cl)C(C)C  
**Mol. weight [g/mol]:** 361.26

## Physical Properties

Property code	Value	Unit	Source
gf	-313.61	kJ/mol	Joback Method
hf	-717.54	kJ/mol	Joback Method
hfus	36.45	kJ/mol	Joback Method
hvap	82.95	kJ/mol	Joback Method
log10ws	-5.41		Crippen Method
logp	4.903		Crippen Method
mvol	265.990	ml/mol	McGowan Method
pc	1591.08	kPa	Joback Method
rinpol	2407.00		NIST Webbook
rinpol	2407.00		NIST Webbook
tb	851.12	K	Joback Method
tc	1068.00	K	Joback Method
tf	491.97	K	Joback Method
vc	1.008	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	756.93	J/molxK	851.12	Joback Method
cpg	770.37	J/molxK	887.27	Joback Method
cpg	782.66	J/molxK	923.41	Joback Method
cpg	793.81	J/molxK	959.56	Joback Method
cpg	803.85	J/molxK	995.71	Joback Method
cpg	812.77	J/molxK	1031.86	Joback Method
cpg	820.61	J/molxK	1068.00	Joback Method
dvisc	0.0006576	Paxs	491.97	Joback Method

dvisc	0.0003368	Paxs	551.83	Joback Method
dvisc	0.0001966	Paxs	611.69	Joback Method
dvisc	0.0001264	Paxs	671.54	Joback Method
dvisc	0.0000873	Paxs	731.40	Joback Method
dvisc	0.0000638	Paxs	791.26	Joback Method
dvisc	0.0000487	Paxs	851.12	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390512&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390512&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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