

# bis-(4-Nitrophenyl)-hydroxyacetic acid, methyl ester

Other names:	para,para'-«alpha»-Hydroxydinitrodiphenylacetic acid, methyl ester
Inchi:	InChI=1S/C15H12N2O7/c1-24-14(18)15(19,10-2-6-12(7-3-10)16(20)21)11-4-8-13(9-5-11)
InchiKey:	NLIGJKPJUSALUSB-UHFFFAOYSA-N
Formula:	C15H12N2O7
SMILES:	COC(=O)C(O)(c1ccc([N+](=O)[O-])cc1)c1ccc([N+](=O)[O-])cc1
Mol. weight [g/mol]:	332.26

## Physical Properties

Property code	Value	Unit	Source
gf	-15.82	kJ/mol	Joback Method
hf	-330.11	kJ/mol	Joback Method
hfus	44.09	kJ/mol	Joback Method
hvap	112.58	kJ/mol	Joback Method
log10ws	-4.01		Crippen Method
logp	1.912		Crippen Method
mcvol	222.840	ml/mol	McGowan Method
pc	2937.70	kPa	Joback Method
rinpol	2750.00		NIST Webbook
rinpol	2750.00		NIST Webbook
rinpol	2732.00		NIST Webbook
rinpol	2732.00		NIST Webbook
rinpol	2732.00		NIST Webbook
tb	1074.84	K	Joback Method
tc	1340.57	K	Joback Method
tf	759.31	K	Joback Method
vc	0.856	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	683.54	J/molxK	1074.84	Joback Method
cpg	690.24	J/molxK	1119.13	Joback Method
cpg	696.19	J/molxK	1163.42	Joback Method
cpg	701.51	J/molxK	1207.71	Joback Method

cpg	706.32	J/mol×K	1252.00	Joback Method
cpg	710.72	J/mol×K	1296.29	Joback Method
cpg	714.85	J/mol×K	1340.57	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R190075&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R190075&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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