

Benzeneacetic acid, 4-nitro-, 2-methyloct-5-yn-4-yl ester

Inchi: InChI=1S/C17H21NO4/c1-4-5-6-16(11-13(2)3)22-17(19)12-14-7-9-15(10-8-14)18(20)21/
InchiKey: ASAYLLCDIXXUSW-UHFFFAOYSA-N
Formula: C17H21NO4
SMILES: CCC#CC(CC(C)C)OC(=O)Cc1ccc([N+](=O)[O-])cc1
Mol. weight [g/mol]: 303.35

Physical Properties

Property code	Value	Unit	Source
gf	194.59	kJ/mol	Joback Method
hf	-162.97	kJ/mol	Joback Method
hfus	43.66	kJ/mol	Joback Method
hvap	83.50	kJ/mol	Joback Method
log10ws	-5.22		Crippen Method
logp	3.509		Crippen Method
mvol	242.890	ml/mol	McGowan Method
pc	1890.36	kPa	Joback Method
rinpol	2236.00		NIST Webbook
rinpol	2236.00		NIST Webbook
tb	856.27	K	Joback Method
tc	1095.10	K	Joback Method
tf	612.16	K	Joback Method
vc	0.935	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	721.89	J/mol×K	856.27	Joback Method
cpg	736.38	J/mol×K	896.07	Joback Method
cpg	749.64	J/mol×K	935.88	Joback Method
cpg	761.71	J/mol×K	975.68	Joback Method
cpg	772.63	J/mol×K	1015.49	Joback Method
cpg	782.46	J/mol×K	1055.29	Joback Method
cpg	791.24	J/mol×K	1095.10	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U406979&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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