

# Isophthalic acid, propyl 2,4,6-trichlorophenyl ester

Inchi:	InChI=1S/C17H13Cl3O4/c1-2-6-23-16(21)10-4-3-5-11(7-10)17(22)24-15-13(19)8-12(18)3
InchiKey:	CANZSVFEHFESDK-UHFFFAOYSA-N
Formula:	C17H13Cl3O4
SMILES:	CCCOC(=O)c1cccc(C(=O)Oc2c(Cl)cc(Cl)cc2Cl)c1
Mol. weight [g/mol]:	387.64

## Physical Properties

Property code	Value	Unit	Source
gf	-225.07	kJ/mol	Joback Method
hf	-503.85	kJ/mol	Joback Method
hfus	44.48	kJ/mol	Joback Method
hvap	92.10	kJ/mol	Joback Method
log10ws	-6.69		Crippen Method
logp	5.433		Crippen Method
mcvol	254.470	ml/mol	McGowan Method
pc	1945.79	kPa	Joback Method
rinpol	2771.00		NIST Webbook
rinpol	2771.00		NIST Webbook
tb	926.51	K	Joback Method
tc	1168.45	K	Joback Method
tf	618.35	K	Joback Method
vc	0.967	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	670.85	J/molxK	926.51	Joback Method
cpg	705.65	J/molxK	1128.13	Joback Method
cpg	701.07	J/molxK	1087.81	Joback Method
cpg	695.31	J/molxK	1047.48	Joback Method
cpg	688.37	J/molxK	1007.16	Joback Method
cpg	680.22	J/molxK	966.83	Joback Method
cpg	709.08	J/molxK	1168.45	Joback Method
dvisc	0.0000583	Paxs	926.51	Joback Method

dvisc	0.0000706	Paxs	875.15	Joback Method
dvisc	0.0000876	Paxs	823.79	Joback Method
dvisc	0.0001118	Paxs	772.43	Joback Method
dvisc	0.0001478	Paxs	721.07	Joback Method
dvisc	0.0002038	Paxs	669.71	Joback Method
dvisc	0.0002965	Paxs	618.35	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U344566&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U344566&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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