

# Phthalic acid, hexadecyl 2-trifluoromethylbenzyl ester

<b>Other names:</b>	Phthalic acid, hexadecyl 2-trifluorobenzyl ester
<b>Inchi:</b>	InChI=1S/C32H43F3O4/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-19-24-38-30(36)27-21-16-17
<b>InchiKey:</b>	PKZIANHZQZOTMI-UHFFFAOYSA-N
<b>Formula:</b>	C32H43F3O4
<b>SMILES:</b>	CCCCCCCCCCCCCCCCOC(=O)c1ccccc1C(=O)OCc1ccccc1C(F)(F)F
<b>Mol. weight [g/mol]:</b>	548.68

## Physical Properties

Property code	Value	Unit	Source
gf	-625.31	kJ/mol	Joback Method
hf	-1340.37	kJ/mol	Joback Method
hfus	73.34	kJ/mol	Joback Method
hvap	107.27	kJ/mol	Joback Method
log10ws	-11.45		Crippen Method
logp	9.701		Crippen Method
mcvol	434.410	ml/mol	McGowan Method
pc	737.62	kPa	Joback Method
rinpol	3521.00		NIST Webbook
rinpol	3521.00		NIST Webbook
tb	1142.04	K	Joback Method
tc	1417.83	K	Joback Method
tf	676.79	K	Joback Method
vc	1.702	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1528.36	J/molxK	1142.04	Joback Method
cpg	1544.58	J/molxK	1188.00	Joback Method
cpg	1559.01	J/molxK	1233.97	Joback Method
cpg	1571.84	J/molxK	1279.93	Joback Method
cpg	1583.26	J/molxK	1325.90	Joback Method
cpg	1593.47	J/molxK	1371.86	Joback Method
cpg	1602.66	J/molxK	1417.83	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U377831&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U377831&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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