

# 1,2-Benzenediol, 3,4,5,6-tetrachloro-

<b>Other names:</b>	Pyrocatechol, tetrachloro- Tetrachloro-1,2-benzenediol Tetrachlorocatechol Tetrachloropyrocatechol 3,4,5,6-Tetrachloro-1,2-benzenediol Tetrachlorpyrokatechin Tetrachlorpyrokatechol
<b>Inchi:</b>	InChI=1S/C6H2Cl4O2/c7-1-2(8)4(10)6(12)5(11)3(1)9/h11-12H
<b>InchiKey:</b>	RRBMVWQICIXSEO-UHFFFAOYSA-N
<b>Formula:</b>	C6H2Cl4O2
<b>SMILES:</b>	Oc1c(O)c(Cl)c(Cl)c(Cl)c1Cl
<b>Mol. weight [g/mol]:</b>	247.89
<b>CAS:</b>	1198-55-6

## Physical Properties

Property code	Value	Unit	Source
gf	-273.80	kJ/mol	Joback Method
hf	-382.63	kJ/mol	Joback Method
hfus	32.52	kJ/mol	Joback Method
hvap	76.78	kJ/mol	Joback Method
log10ws	-3.31		Crippen Method
logp	3.711		Crippen Method
mcvol	132.340	ml/mol	McGowan Method
pc	5422.51	kPa	Joback Method
tb	689.26	K	Joback Method
tc	953.67	K	Joback Method
tf	564.48	K	Joback Method
vc	0.392	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	247.27	J/mol×K	689.26	Joback Method
cpg	251.42	J/mol×K	733.33	Joback Method

cpg	255.42	J/molxK	777.40	Joback Method
cpg	259.44	J/molxK	821.47	Joback Method
cpg	263.62	J/molxK	865.54	Joback Method
cpg	268.15	J/molxK	909.61	Joback Method
cpg	273.16	J/molxK	953.67	Joback Method
dvisc	0.0000238	Paxs	564.48	Joback Method
dvisc	0.0000158	Paxs	585.28	Joback Method
dvisc	0.0000108	Paxs	606.07	Joback Method
dvisc	0.0000076	Paxs	626.87	Joback Method
dvisc	0.0000055	Paxs	647.67	Joback Method
dvisc	0.0000040	Paxs	668.46	Joback Method
dvisc	0.0000030	Paxs	689.26	Joback Method
hvapt	77.90	kJ/mol	308.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1198556&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1198556&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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