

Diglycolic acid, 2-chloro-6-fluorophenyl isoheptyl ester

Inchi:	InChI=1S/C16H20ClFO5/c1-11(2)5-4-8-22-14(19)9-21-10-15(20)23-16-12(17)6-3-7-13(18)
InchiKey:	UFFHXJOUVHZDAK-UHFFFAOYSA-N
Formula:	C16H20ClFO5
SMILES:	CC(C)CCCOC(=O)COCC(=O)Oc1c(F)cccc1Cl
Mol. weight [g/mol]:	346.78

Physical Properties

Property code	Value	Unit	Source
gf	-605.03	kJ/mol	Joback Method
hf	-998.93	kJ/mol	Joback Method
hfus	40.97	kJ/mol	Joback Method
hvap	78.71	kJ/mol	Joback Method
log10ws	-3.86		Crippen Method
logp	3.380		Crippen Method
mvol	247.300	ml/mol	McGowan Method
pc	1663.26	kPa	Joback Method
rinpol	2717.00		NIST Webbook
rinpol	2717.00		NIST Webbook
tb	813.38	K	Joback Method
tc	1016.16	K	Joback Method
tf	503.60	K	Joback Method
vc	0.951	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	709.51	J/molxK	813.38	Joback Method
cpg	722.71	J/molxK	847.18	Joback Method
cpg	734.89	J/molxK	880.97	Joback Method
cpg	746.04	J/molxK	914.77	Joback Method
cpg	756.17	J/molxK	948.56	Joback Method
cpg	765.26	J/molxK	982.36	Joback Method
cpg	773.32	J/molxK	1016.16	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U381943&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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