

Pimelic acid, isobutyl 2-nitrophenyl ester

Inchi:	InChI=1S/C17H23NO6/c1-13(2)12-23-16(19)10-4-3-5-11-17(20)24-15-9-7-6-8-14(15)18(
InchiKey:	FVHAMDGYGLAOMH-UHFFFAOYSA-N
Formula:	C17H23NO6
SMILES:	CC(C)COC(=O)CCCCC(=O)Oc1ccccc1[N+](=O)[O-]
Mol. weight [g/mol]:	337.37

Physical Properties

Property code	Value	Unit	Source
gf	-239.69	kJ/mol	Joback Method
hf	-674.79	kJ/mol	Joback Method
hfus	46.85	kJ/mol	Joback Method
hvap	90.89	kJ/mol	Joback Method
log10ws	-4.82		Crippen Method
logp	3.650		Crippen Method
mcvol	258.930	ml/mol	McGowan Method
pc	1704.71	kPa	Joback Method
rinpola	2483.00		NIST Webbook
rinpola	2483.00		NIST Webbook
tb	924.00	K	Joback Method
tc	1148.18	K	Joback Method
tf	593.22	K	Joback Method
vc	1.004	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	812.88	J/molxK	924.00	Joback Method
cpg	825.20	J/molxK	961.36	Joback Method
cpg	836.26	J/molxK	998.73	Joback Method
cpg	846.10	J/molxK	1036.09	Joback Method
cpg	854.75	J/molxK	1073.45	Joback Method
cpg	862.23	J/molxK	1110.82	Joback Method
cpg	868.56	J/molxK	1148.18	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U416743&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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