

# Succinic acid, di(2-(2-chlorophenoxy)ethyl) ester

Inchi:	InChI=1S/C20H20Cl2O6/c21-15-5-1-3-7-17(15)25-11-13-27-19(23)9-10-20(24)28-14-12-
InchiKey:	QNKNESZJDWBOEA-UHFFFAOYSA-N
Formula:	C20H20Cl2O6
SMILES:	O=C(CCC(=O)OCCOc1ccccc1Cl)OCCOc1ccccc1Cl
Mol. weight [g/mol]:	427.27

## Physical Properties

Property code	Value	Unit	Source
gf	-378.62	kJ/mol	Joback Method
hf	-791.53	kJ/mol	Joback Method
hfus	51.20	kJ/mol	Joback Method
hvap	97.89	kJ/mol	Joback Method
log10ws	-4.97		Crippen Method
logp	4.318		Crippen Method
mcvol	296.240	ml/mol	McGowan Method
pc	1572.21	kPa	Joback Method
rinpol	3223.00		NIST Webbook
rinpol	3223.00		NIST Webbook
tb	992.60	K	Joback Method
tc	1225.33	K	Joback Method
tf	641.66	K	Joback Method
vc	1.121	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	874.69	J/molxK	992.60	Joback Method
cpg	905.76	J/molxK	1186.54	Joback Method
cpg	902.69	J/molxK	1147.75	Joback Method
cpg	898.06	J/molxK	1108.97	Joback Method
cpg	891.84	J/molxK	1070.18	Joback Method
cpg	884.05	J/molxK	1031.39	Joback Method
cpg	907.26	J/molxK	1225.33	Joback Method
dvisc	0.0000232	Paxs	992.60	Joback Method

dvisc	0.0000289	Paxs	934.11	Joback Method
dvisc	0.0000371	Paxs	875.62	Joback Method
dvisc	0.0000493	Paxs	817.13	Joback Method
dvisc	0.0000684	Paxs	758.64	Joback Method
dvisc	0.0001004	Paxs	700.15	Joback Method
dvisc	0.0001580	Paxs	641.66	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U381545&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U381545&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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