

2,5-Di(trifluoromethyl)benzoic acid, 3-fluorophenyl ester

Inchi:	InChI=1S/C15H7F7O2/c16-9-2-1-3-10(7-9)24-13(23)11-6-8(14(17,18)19)4-5-12(11)15(20)
InchiKey:	LWBRNIAJKIFWCD-UHFFFAOYSA-N
Formula:	C15H7F7O2
SMILES:	O=C(Oc1cccc(F)c1)c1cc(C(F)(F)F)ccc1C(F)(F)F
Mol. weight [g/mol]:	352.20

Physical Properties

Property code	Value	Unit	Source
gf	-1320.56	kJ/mol	Joback Method
hf	-1549.35	kJ/mol	Joback Method
hfus	31.04	kJ/mol	Joback Method
hvap	56.37	kJ/mol	Joback Method
log10ws	-6.10		Crippen Method
logp	5.082		Crippen Method
mvol	194.520	ml/mol	McGowan Method
pc	1937.24	kPa	Joback Method
rinpol	1528.00		NIST Webbook
rinpol	1528.00		NIST Webbook
tb	675.62	K	Joback Method
tc	873.27	K	Joback Method
tf	430.34	K	Joback Method
vc	0.787	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	525.89	J/mol×K	675.62	Joback Method
cpg	537.54	J/mol×K	708.56	Joback Method
cpg	548.27	J/mol×K	741.50	Joback Method
cpg	558.13	J/mol×K	774.44	Joback Method
cpg	567.19	J/mol×K	807.39	Joback Method
cpg	575.50	J/mol×K	840.33	Joback Method
cpg	583.12	J/mol×K	873.27	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U357366&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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