

# Fumaric acid, 3,5-dimethylphenyl 2,2-dichloroethyl ester

<b>Inchi:</b>	InChI=1S/C14H14Cl2O4/c1-9-5-10(2)7-11(6-9)20-14(18)4-3-13(17)19-8-12(15)16/h3-7,1
<b>InchiKey:</b>	NPGWRYIYWRULSV-ONEGZZNKSA-N
<b>Formula:</b>	C14H14Cl2O4
<b>SMILES:</b>	Cc1cc(C)cc(OC(=O)C=CC(=O)OCC(Cl)Cl)c1
<b>Mol. weight [g/mol]:</b>	317.17

## Physical Properties

Property code	Value	Unit	Source
gf	-253.77	kJ/mol	Joback Method
hf	-527.84	kJ/mol	Joback Method
hfus	35.93	kJ/mol	Joback Method
hvap	77.01	kJ/mol	Joback Method
log10ws	-4.04		Crippen Method
logp	3.112		Crippen Method
mvol	219.420	ml/mol	McGowan Method
pc	2098.42	kPa	Joback Method
rinpol	2198.00		NIST Webbook
rinpol	2198.00		NIST Webbook
tb	787.52	K	Joback Method
tc	1011.27	K	Joback Method
tf	483.08	K	Joback Method
vc	0.832	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	563.75	J/molxK	787.52	Joback Method
cpg	575.48	J/molxK	824.81	Joback Method
cpg	586.29	J/molxK	862.10	Joback Method
cpg	596.19	J/molxK	899.40	Joback Method
cpg	605.20	J/molxK	936.69	Joback Method
cpg	613.35	J/molxK	973.98	Joback Method
cpg	620.65	J/molxK	1011.27	Joback Method
dvisc	0.0006394	Paxs	483.08	Joback Method

dvisc	0.0003794	Paxs	533.82	Joback Method
dvisc	0.0002464	Paxs	584.56	Joback Method
dvisc	0.0001715	Paxs	635.30	Joback Method
dvisc	0.0001259	Paxs	686.04	Joback Method
dvisc	0.0000965	Paxs	736.78	Joback Method
dvisc	0.0000765	Paxs	787.52	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U405738&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U405738&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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