

# 1,2-Cyclohexanedicarboxylic acid, di(2-chloroethyl) ester

**Inchi:** InChI=1S/C12H18Cl2O4/c13-5-7-17-11(15)9-3-1-2-4-10(9)12(16)18-8-6-14/h9-10H,1-8H  
**InchiKey:** IKIDGIJIVUSXNH-UHFFFAOYSA-N  
**Formula:** C12H18Cl2O4  
**SMILES:** O=C(OCCCl)C1CCCCC1C(=O)OCCCl  
**Mol. weight [g/mol]:** 297.18

## Physical Properties

Property code	Value	Unit	Source
gf	-424.80	kJ/mol	Joback Method
hf	-778.11	kJ/mol	Joback Method
hfus	33.71	kJ/mol	Joback Method
hvap	69.51	kJ/mol	Joback Method
log10ws	-2.29		Crippen Method
logp	2.357		Crippen Method
mvol	208.440	ml/mol	McGowan Method
pc	2104.20	kPa	Joback Method
rinpol	2040.00		NIST Webbook
rinpol	2040.00		NIST Webbook
tb	716.28	K	Joback Method
tc	927.94	K	Joback Method
tf	432.30	K	Joback Method
vc	0.785	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	568.19	J/molxK	716.28	Joback Method
cpg	583.57	J/molxK	751.56	Joback Method
cpg	597.88	J/molxK	786.83	Joback Method
cpg	611.13	J/molxK	822.11	Joback Method
cpg	623.32	J/molxK	857.39	Joback Method
cpg	634.43	J/molxK	892.66	Joback Method
cpg	644.49	J/molxK	927.94	Joback Method
dvisc	0.0014288	Paxs	432.30	Joback Method

dvisc	0.0008269	Paxs	479.63	Joback Method
dvisc	0.0005279	Paxs	526.96	Joback Method
dvisc	0.0003629	Paxs	574.29	Joback Method
dvisc	0.0002641	Paxs	621.62	Joback Method
dvisc	0.0002011	Paxs	668.95	Joback Method
dvisc	0.0001587	Paxs	716.28	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U340054&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U340054&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>c</sub>vol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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