

Diethylmalonic acid, dipentafluorobenzyl ester

Inchi: InChI=1S/C21H14F10O4/c1-3-21(4-2,19(32)34-5-7-9(22)13(26)17(30)14(27)10(7)23)20(31)2
InchiKey: AXWJURRSJIOQFZ-UHFFFAOYSA-N
Formula: C21H14F10O4
SMILES: CCC(CC)(C(=O)OCc1c(F)c(F)c(F)c(F)c1F)C(=O)OCc1c(F)c(F)c(F)c(F)c1F
Mol. weight [g/mol]: 520.32

Physical Properties

Property code	Value	Unit	Source
gf	-2158.64	kJ/mol	Joback Method
hf	-2577.86	kJ/mol	Joback Method
hfus	63.30	kJ/mol	Joback Method
hvap	82.36	kJ/mol	Joback Method
log10ws	-8.61		Crippen Method
logp	5.671		Crippen Method
mcvol	291.810	ml/mol	McGowan Method
pc	1067.27	kPa	Joback Method
rinpol	2026.00		NIST Webbook
rinpol	2026.00		NIST Webbook
tb	925.09	K	Joback Method
tc	1133.14	K	Joback Method
tf	657.11	K	Joback Method
vc	1.212	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	900.68	J/molxK	925.09	Joback Method
cpg	911.43	J/molxK	959.76	Joback Method
cpg	921.05	J/molxK	994.44	Joback Method
cpg	929.56	J/molxK	1029.11	Joback Method
cpg	936.96	J/molxK	1063.79	Joback Method
cpg	943.26	J/molxK	1098.46	Joback Method
cpg	948.47	J/molxK	1133.14	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U370006&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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