

# (3-Chloro-4-cyclohexylmethoxy-phenyl)-propionic acid, methyl ester

Inchi: nCl=18/C17H23ClO3/c1-20-17(19)10-8-13-7-9-16(15(18)11-13)21-12-14-5-3-2-4-6-14/  
InchiKey: RTRDUUMGKSDSRB-UHFFFAOYSA-N

Formula: C17H23ClO3

SMILES: COC(=O)CCc1ccc(OCC2CCCCC2)c(Cl)c1

Mol. weight [g/mol]: 310.82

## Physical Properties

Property code	Value	Unit	Source
gf	-140.99	kJ/mol	Joback Method
hf	-519.06	kJ/mol	Joback Method
hfus	33.06	kJ/mol	Joback Method
hvap	73.42	kJ/mol	Joback Method
log10ws	-4.94		Crippen Method
logp	4.405		Crippen Method
mcvol	241.320	ml/mol	McGowan Method
pc	1807.70	kPa	Joback Method
rinpol	2322.60		NIST Webbook
rinpol	2322.60		NIST Webbook
tb	780.69	K	Joback Method
tc	1004.95	K	Joback Method
tf	464.50	K	Joback Method
vc	0.903	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	707.39	J/molxK	780.69	Joback Method
cpg	724.96	J/molxK	818.07	Joback Method
cpg	741.15	J/molxK	855.44	Joback Method
cpg	755.98	J/molxK	892.82	Joback Method
cpg	769.47	J/molxK	930.20	Joback Method
cpg	781.63	J/molxK	967.57	Joback Method
cpg	792.50	J/molxK	1004.95	Joback Method
dvisc	0.0007724	Paxs	464.50	Joback Method

dvisc	0.0004340	Paxs	517.20	Joback Method
dvisc	0.0002713	Paxs	569.90	Joback Method
dvisc	0.0001836	Paxs	622.60	Joback Method
dvisc	0.0001321	Paxs	675.29	Joback Method
dvisc	0.0000997	Paxs	727.99	Joback Method
dvisc	0.0000781	Paxs	780.69	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R157989&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R157989&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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