

# Succinic acid, 2,3-dichlorophenyl but-2-en-1-yl ester

**Inchi:** InChI=1S/C14H14Cl2O4/c1-2-3-9-19-12(17)7-8-13(18)20-11-6-4-5-10(15)14(11)16/h2-6H  
**InchiKey:** AFXHVBQTVTVWCK-NSCUHMNNSA-N  
**Formula:** C14H14Cl2O4  
**SMILES:** CC=CCOC(=O)CCC(=O)Oc1cccc(Cl)c1Cl  
**Mol. weight [g/mol]:** 317.17

## Physical Properties

Property code	Value	Unit	Source
gf	-251.33	kJ/mol	Joback Method
hf	-522.56	kJ/mol	Joback Method
hfus	39.45	kJ/mol	Joback Method
hvap	77.40	kJ/mol	Joback Method
log10ws	-4.38		Crippen Method
logp	3.798		Crippen Method
mcvol	219.420	ml/mol	McGowan Method
pc	2083.12	kPa	Joback Method
rinpola	2250.00		NIST Webbook
rinpola	2250.00		NIST Webbook
tb	787.96	K	Joback Method
tc	1008.22	K	Joback Method
tf	498.08	K	Joback Method
vc	0.838	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	563.14	J/molxK	787.96	Joback Method
cpg	574.70	J/molxK	824.67	Joback Method
cpg	585.37	J/molxK	861.38	Joback Method
cpg	595.17	J/molxK	898.09	Joback Method
cpg	604.12	J/molxK	934.80	Joback Method
cpg	612.24	J/molxK	971.51	Joback Method
cpg	619.56	J/molxK	1008.22	Joback Method
dvisc	0.0005743	Paxs	498.08	Joback Method

dvisc	0.0003610	Paxs	546.39	Joback Method
dvisc	0.0002447	Paxs	594.71	Joback Method
dvisc	0.0001759	Paxs	643.02	Joback Method
dvisc	0.0001324	Paxs	691.33	Joback Method
dvisc	0.0001034	Paxs	739.65	Joback Method
dvisc	0.0000832	Paxs	787.96	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391229&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391229&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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