

Fumaric acid, isobutyl 2,3,5-trichlorophenyl ester

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|----------------------|---|
| Inchi: | InChI=1S/C14H13Cl3O4/c1-8(2)7-20-12(18)3-4-13(19)21-11-6-9(15)5-10(16)14(11)17/h3 |
| InchiKey: | LEPGKNXBRAIHJE-ONEGZZNKSA-N |
| Formula: | C14H13Cl3O4 |
| SMILES: | CC(C)COC(=O)C=CC(=O)Oc1cc(Cl)cc(Cl)c1Cl |
| Mol. weight [g/mol]: | 351.61 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -275.33 | kJ/mol | Joback Method |
| hf | -555.05 | kJ/mol | Joback Method |
| hfus | 39.73 | kJ/mol | Joback Method |
| hvap | 82.06 | kJ/mol | Joback Method |
| log10ws | -4.83 | | Crippen Method |
| logp | 4.308 | | Crippen Method |
| mvol | 231.660 | ml/mol | McGowan Method |
| pc | 2001.91 | kPa | Joback Method |
| rinpol | 2268.00 | | NIST Webbook |
| rinpol | 2268.00 | | NIST Webbook |
| tb | 829.93 | K | Joback Method |
| tc | 1057.31 | K | Joback Method |
| tf | 525.52 | K | Joback Method |
| vc | 0.880 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 583.51 | J/molxK | 829.93 | Joback Method |
| cpg | 594.06 | J/molxK | 867.83 | Joback Method |
| cpg | 603.69 | J/molxK | 905.72 | Joback Method |
| cpg | 612.41 | J/molxK | 943.62 | Joback Method |
| cpg | 620.25 | J/molxK | 981.52 | Joback Method |
| cpg | 627.23 | J/molxK | 1019.41 | Joback Method |
| cpg | 633.35 | J/molxK | 1057.31 | Joback Method |
| dvisc | 0.0004680 | Paxs | 525.52 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0002941 | Paxs | 576.25 | Joback Method |
| dvisc | 0.0001993 | Paxs | 626.99 | Joback Method |
| dvisc | 0.0001431 | Paxs | 677.72 | Joback Method |
| dvisc | 0.0001076 | Paxs | 728.46 | Joback Method |
| dvisc | 0.0000840 | Paxs | 779.19 | Joback Method |
| dvisc | 0.0000676 | Paxs | 829.93 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U348141&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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