

# 1-chloropropyl trichloroacetate

<b>Other names:</b>	1-Propanol, 1-chloro, trichloroacetate
<b>Inchi:</b>	InChI=1S/C5H6Cl4O2/c1-2-3(6)11-4(10)5(7,8)9/h3H,2H2,1H3
<b>InchiKey:</b>	IABUENOUASULAK-UHFFFAOYSA-N
<b>Formula:</b>	C5H6Cl4O2
<b>SMILES:</b>	CCC(Cl)OC(=O)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	239.91

## Physical Properties

Property code	Value	Unit	Source
gf	-290.02	kJ/mol	Joback Method
hf	-468.32	kJ/mol	Joback Method
hfus	17.34	kJ/mol	Joback Method
hvap	51.74	kJ/mol	Joback Method
log10ws	-3.10		Crippen Method
logp	2.875		Crippen Method
mcvol	137.710	ml/mol	McGowan Method
pc	3181.14	kPa	Joback Method
rinpol	1129.00		NIST Webbook
rinpol	1123.00		NIST Webbook
rinpol	1117.00		NIST Webbook
rinpol	1123.00		NIST Webbook
rinpol	1163.00		NIST Webbook
rinpol	1154.00		NIST Webbook
ripol	1598.00		NIST Webbook
ripol	1611.00		NIST Webbook
ripol	1628.00		NIST Webbook
ripol	1590.00		NIST Webbook
ripol	1590.00		NIST Webbook
ripol	1591.00		NIST Webbook
tb	536.14	K	Joback Method
tc	757.08	K	Joback Method
tf	325.37	K	Joback Method
vc	0.518	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	255.96	J/molxK	536.14	Joback Method
cpg	263.88	J/molxK	572.96	Joback Method
cpg	271.22	J/molxK	609.79	Joback Method
cpg	278.00	J/molxK	646.61	Joback Method
cpg	284.24	J/molxK	683.43	Joback Method
cpg	289.98	J/molxK	720.26	Joback Method
cpg	295.24	J/molxK	757.08	Joback Method
dvisc	0.0035419	Paxs	325.37	Joback Method
dvisc	0.0018810	Paxs	360.50	Joback Method
dvisc	0.0011178	Paxs	395.63	Joback Method
dvisc	0.0007231	Paxs	430.75	Joback Method
dvisc	0.0004995	Paxs	465.88	Joback Method
dvisc	0.0003634	Paxs	501.01	Joback Method
dvisc	0.0002757	Paxs	536.14	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R112535&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R112535&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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