

# Adipic acid, isohexyl 2,3,5,6-tetrachlorophenyl ester

Inchi:	InChI=1S/C18H22Cl4O4/c1-11(2)6-5-9-25-14(23)7-3-4-8-15(24)26-18-16(21)12(19)10-13
InchiKey:	OOWKGGQXZLINJFL-UHFFFAOYSA-N
Formula:	C18H22Cl4O4
SMILES:	CC(C)CCCOC(=O)CCCCC(=O)Oc1c(Cl)c(Cl)cc(Cl)c1Cl
Mol. weight [g/mol]:	444.18

## Physical Properties

Property code	Value	Unit	Source
gf	-343.43	kJ/mol	Joback Method
hf	-782.04	kJ/mol	Joback Method
hfus	53.70	kJ/mol	Joback Method
hvap	96.05	kJ/mol	Joback Method
log10ws	-7.33		Crippen Method
logp	6.745		Crippen Method
mcvol	304.560	ml/mol	McGowan Method
pc	1348.67	kPa	Joback Method
rinpol	2839.00		NIST Webbook
rinpol	2839.00		NIST Webbook
tb	959.70	K	Joback Method
tc	1182.94	K	Joback Method
tf	618.12	K	Joback Method
vc	1.173	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	854.34	J/molxK	959.70	Joback Method
cpg	865.10	J/molxK	996.91	Joback Method
cpg	874.65	J/molxK	1034.11	Joback Method
cpg	883.01	J/molxK	1071.32	Joback Method
cpg	890.19	J/molxK	1108.52	Joback Method
cpg	896.21	J/molxK	1145.73	Joback Method
cpg	901.06	J/molxK	1182.94	Joback Method
dvisc	0.0002506	Paxs	618.12	Joback Method

dvisc	0.0001599	Paxs	675.05	Joback Method
dvisc	0.0001094	Paxs	731.98	Joback Method
dvisc	0.0000791	Paxs	788.91	Joback Method
dvisc	0.0000597	Paxs	845.84	Joback Method
dvisc	0.0000467	Paxs	902.77	Joback Method
dvisc	0.0000376	Paxs	959.70	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353950&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353950&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rin<sub>pol</sub>:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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