

# 4-Aminobenzoic acid, N,O-bis(pentafluoropropionyl)-

**Inchi:** InChI=1S/C13H5F10NO4/c14-10(15,12(18,19)20)8(26)24-6-3-1-5(2-4-6)7(25)28-9(27)11  
**InchiKey:** ZHCVZWKXNNLIDU-UHFFFAOYSA-N  
**Formula:** C13H5F10NO4  
**SMILES:** O=C(OC(=O)C(F)(F)C(F)(F)F)c1ccc(NC(=O)C(F)(F)C(F)(F)F)cc1  
**Mol. weight [g/mol]:** 429.17

## Physical Properties

Property code	Value	Unit	Source
gf	-2177.75	kJ/mol	Joback Method
hf	-2499.18	kJ/mol	Joback Method
hfus	35.31	kJ/mol	Joback Method
hvap	63.20	kJ/mol	Joback Method
log10ws	-4.90		Crippen Method
logp	3.704		Crippen Method
mcvol	208.530	ml/mol	McGowan Method
pc	1827.85	kPa	Joback Method
rinpol	1475.00		NIST Webbook
rinpol	1475.00		NIST Webbook
tb	742.48	K	Joback Method
tc	926.16	K	Joback Method
tf	515.47	K	Joback Method
vc	0.863	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	615.87	J/mol×K	742.48	Joback Method
cpg	624.32	J/mol×K	773.09	Joback Method
cpg	631.99	J/mol×K	803.71	Joback Method
cpg	638.95	J/mol×K	834.32	Joback Method
cpg	645.28	J/mol×K	864.93	Joback Method
cpg	651.05	J/mol×K	895.55	Joback Method
cpg	656.33	J/mol×K	926.16	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U375076&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U375076&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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