

«alpha»-Hydroxybehenic acid, HFB-Me

Inchi:	InChI=1S/C27H45F7O4/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22(23(3
InchiKey:	JBLKXFRSPSLDPU-UHFFFAOYSA-N
Formula:	C27H45F7O4
SMILES:	CCCCCCCCCCCCCCCCCCCC(OC(=O)C(F)(F)C(F)(F)C(F)(F)C(F)(F)C(=O)OC
Mol. weight [g/mol]:	566.63

Physical Properties

Property code	Value	Unit	Source
gf	-1648.97	kJ/mol	Joback Method
hf	-2494.51	kJ/mol	Joback Method
hfus	67.05	kJ/mol	Joback Method
hvap	84.01	kJ/mol	Joback Method
log10ws	-10.25		Crippen Method
logp	9.336		Crippen Method
mvol	418.560	ml/mol	McGowan Method
pc	635.12	kPa	Joback Method
rinpol	2564.00		NIST Webbook
rinpol	2564.00		NIST Webbook
tb	954.50	K	Joback Method
tc	1192.73	K	Joback Method
tf	534.76	K	Joback Method
vc	1.683	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1456.79	J/mol×K	954.50	Joback Method
cpg	1478.70	J/mol×K	994.20	Joback Method
cpg	1498.99	J/mol×K	1033.91	Joback Method
cpg	1517.81	J/mol×K	1073.61	Joback Method
cpg	1535.33	J/mol×K	1113.32	Joback Method
cpg	1551.72	J/mol×K	1153.02	Joback Method
cpg	1567.14	J/mol×K	1192.73	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R134228&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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