

# Glutaric acid, ethyl pentachlorophenyl ester

**Inchi:** InChI=1S/C13H11Cl5O4/c1-2-21-6(19)4-3-5-7(20)22-13-11(17)9(15)8(14)10(16)12(13)18  
**InchiKey:** BTUGUQUIQVXBBT-UHFFFAOYSA-N  
**Formula:** C13H11Cl5O4  
**SMILES:** CCOC(=O)CCCC(=O)Oc1c(Cl)c(Cl)c(Cl)c(Cl)c1Cl  
**Mol. weight [g/mol]:** 408.49

## Physical Properties

Property code	Value	Unit	Source
gf	-404.65	kJ/mol	Joback Method
hf	-700.77	kJ/mol	Joback Method
hfus	48.08	kJ/mol	Joback Method
hvap	90.36	kJ/mol	Joback Method
log10ws	-6.17		Crippen Method
logp	5.592		Crippen Method
mvol	246.350	ml/mol	McGowan Method
pc	1885.44	kPa	Joback Method
rinpol	2589.00		NIST Webbook
rinpol	2589.00		NIST Webbook
tb	888.15	K	Joback Method
tc	1116.03	K	Joback Method
tf	619.21	K	Joback Method
vc	0.949	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	591.13	J/molxK	888.15	Joback Method
cpg	622.74	J/molxK	1078.05	Joback Method
cpg	618.36	J/molxK	1040.07	Joback Method
cpg	613.00	J/molxK	1002.09	Joback Method
cpg	606.66	J/molxK	964.11	Joback Method
cpg	599.37	J/molxK	926.13	Joback Method
cpg	626.12	J/molxK	1116.03	Joback Method
dvisc	0.0000746	Paxs	888.15	Joback Method

dvisc	0.0000887	Paxs	843.33	Joback Method
dvisc	0.0001076	Paxs	798.50	Joback Method
dvisc	0.0001336	Paxs	753.68	Joback Method
dvisc	0.0001705	Paxs	708.86	Joback Method
dvisc	0.0002249	Paxs	664.03	Joback Method
dvisc	0.0003086	Paxs	619.21	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U360251&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U360251&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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