

# Glutaric acid, 4-bromo-2-methoxyphenyl ethyl ester

Inchi:	InChI=1S/C14H17BrO5/c1-3-19-13(16)5-4-6-14(17)20-11-8-7-10(15)9-12(11)18-2/h7-9H
InchiKey:	KXLUJJPWVGXOEI-UHFFFAOYSA-N
Formula:	C14H17BrO5
SMILES:	CCOC(=O)CCCC(=O)Oc1ccc(Br)cc1OC
Mol. weight [g/mol]:	345.19

## Physical Properties

Property code	Value	Unit	Source
gf	-398.37	kJ/mol	Joback Method
hf	-714.19	kJ/mol	Joback Method
hfus	37.33	kJ/mol	Joback Method
hvap	77.52	kJ/mol	Joback Method
log10ws	-4.02		Crippen Method
logp	3.096		Crippen Method
mvol	222.610	ml/mol	McGowan Method
pc	2231.30	kPa	Joback Method
rinpol	2221.00		NIST Webbook
rinpol	2221.00		NIST Webbook
tb	797.52	K	Joback Method
tc	1012.86	K	Joback Method
tf	525.35	K	Joback Method
vc	0.840	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	602.37	J/molxK	797.52	Joback Method
cpg	614.80	J/molxK	833.41	Joback Method
cpg	626.24	J/molxK	869.30	Joback Method
cpg	636.71	J/molxK	905.19	Joback Method
cpg	646.19	J/molxK	941.08	Joback Method
cpg	654.68	J/molxK	976.97	Joback Method
cpg	662.17	J/molxK	1012.86	Joback Method
dvisc	0.0004471	Paxs	525.35	Joback Method

dvisc	0.0002970	Paxs	570.71	Joback Method
dvisc	0.0002096	Paxs	616.07	Joback Method
dvisc	0.0001551	Paxs	661.43	Joback Method
dvisc	0.0001193	Paxs	706.80	Joback Method
dvisc	0.0000948	Paxs	752.16	Joback Method
dvisc	0.0000772	Paxs	797.52	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U393898&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U393898&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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