

# Glutaric acid, 2,2-dichloroethyl 4-methylpent-2-yl ester

<b>Inchi:</b>	InChI=1S/C13H22Cl2O4/c1-9(2)7-10(3)19-13(17)6-4-5-12(16)18-8-11(14)15/h9-11H,4-8
<b>InchiKey:</b>	VGYYIHUSCJTTFJW-UHFFFAOYSA-N
<b>Formula:</b>	C13H22Cl2O4
<b>SMILES:</b>	CC(C)CC(C)OC(=O)CCCC(=O)OCC(Cl)Cl
<b>Mol. weight [g/mol]:</b>	313.22

## Physical Properties

Property code	Value	Unit	Source
gf	-440.44	kJ/mol	Joback Method
hf	-848.57	kJ/mol	Joback Method
hfus	32.82	kJ/mol	Joback Method
hvap	70.45	kJ/mol	Joback Method
log10ws	-3.77		Crippen Method
logp	3.481		Crippen Method
mvol	233.390	ml/mol	McGowan Method
pc	1701.90	kPa	Joback Method
rinpol	1842.00		NIST Webbook
rinpol	1842.00		NIST Webbook
tb	722.96	K	Joback Method
tc	915.97	K	Joback Method
tf	395.43	K	Joback Method
vc	0.891	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	628.19	J/molxK	722.96	Joback Method
cpg	642.33	J/molxK	755.13	Joback Method
cpg	655.64	J/molxK	787.30	Joback Method
cpg	668.12	J/molxK	819.47	Joback Method
cpg	679.79	J/molxK	851.63	Joback Method
cpg	690.64	J/molxK	883.80	Joback Method
cpg	700.70	J/molxK	915.97	Joback Method
dvisc	0.0019317	Paxs	395.43	Joback Method

dvisc	0.0008368	Paxs	450.02	Joback Method
dvisc	0.0004344	Paxs	504.61	Joback Method
dvisc	0.0002563	Paxs	559.20	Joback Method
dvisc	0.0001661	Paxs	613.78	Joback Method
dvisc	0.0001156	Paxs	668.37	Joback Method
dvisc	0.0000849	Paxs	722.96	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U392487&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U392487&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>c</sub>vol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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