

# Succinic acid, 2,4,6-trichlorophenyl neopentyl ester

Inchi:	InChI=1S/C15H17Cl3O4/c1-15(2,3)8-21-12(19)4-5-13(20)22-14-10(17)6-9(16)7-11(14)18
InchiKey:	KJJILITUVBWDOM-UHFFFAOYSA-N
Formula:	C15H17Cl3O4
SMILES:	CC(C)(C)COC(=O)CCC(=O)Oc1c(Cl)cc(Cl)cc1Cl
Mol. weight [g/mol]:	367.65

## Physical Properties

Property code	Value	Unit	Source
gf	-341.85	kJ/mol	Joback Method
hf	-696.38	kJ/mol	Joback Method
hfus	38.23	kJ/mol	Joback Method
hvap	83.42	kJ/mol	Joback Method
log10ws	-5.39		Crippen Method
logp	4.922		Crippen Method
mcvol	250.050	ml/mol	McGowan Method
pc	1777.34	kPa	Joback Method
rinpola	2271.00		NIST Webbook
rinpola	2271.00		NIST Webbook
tb	845.86	K	Joback Method
tc	1069.99	K	Joback Method
tf	559.29	K	Joback Method
vc	0.952	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	666.15	J/molxK	845.86	Joback Method
cpg	713.42	J/molxK	1032.64	Joback Method
cpg	705.86	J/molxK	995.28	Joback Method
cpg	697.38	J/molxK	957.93	Joback Method
cpg	687.95	J/molxK	920.57	Joback Method
cpg	677.55	J/molxK	883.22	Joback Method
cpg	720.09	J/molxK	1069.99	Joback Method
dvisc	0.0000580	Paxs	845.86	Joback Method

dvisc	0.0000722	Paxs	798.10	Joback Method
dvisc	0.0000924	Paxs	750.34	Joback Method
dvisc	0.0001223	Paxs	702.58	Joback Method
dvisc	0.0001687	Paxs	654.81	Joback Method
dvisc	0.0002446	Paxs	607.05	Joback Method
dvisc	0.0003780	Paxs	559.29	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U389591&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U389591&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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